

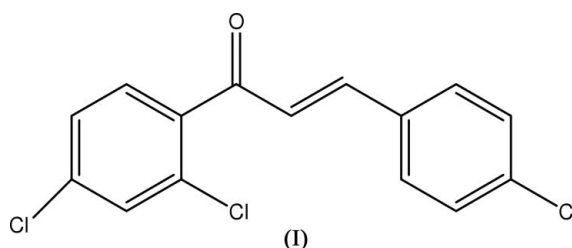
3-(4-Chlorophenyl)-1-(2,4-dichlorophenyl)-  
prop-2-en-1-oneP. S. Patil,<sup>a</sup> Shea-Lin Ng,<sup>b</sup>  
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## Key indicators

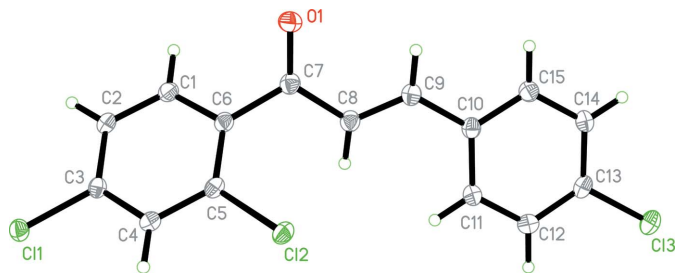
Single-crystal X-ray study  
 $T = 100\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$   
 $R$  factor = 0.049  
 $wR$  factor = 0.137  
Data-to-parameter ratio = 35.7For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.In the title compound,  $\text{C}_{15}\text{H}_9\text{Cl}_3\text{O}$ , the enone unit and the benzene rings are each planar. The molecules are linked *via* weak  $\text{C}-\text{H}\cdots\text{Cl}$  interactions into wave-like chains along the *a* axis and are stacked parallel to the *b* axis.Received 8 March 2006  
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## Comment

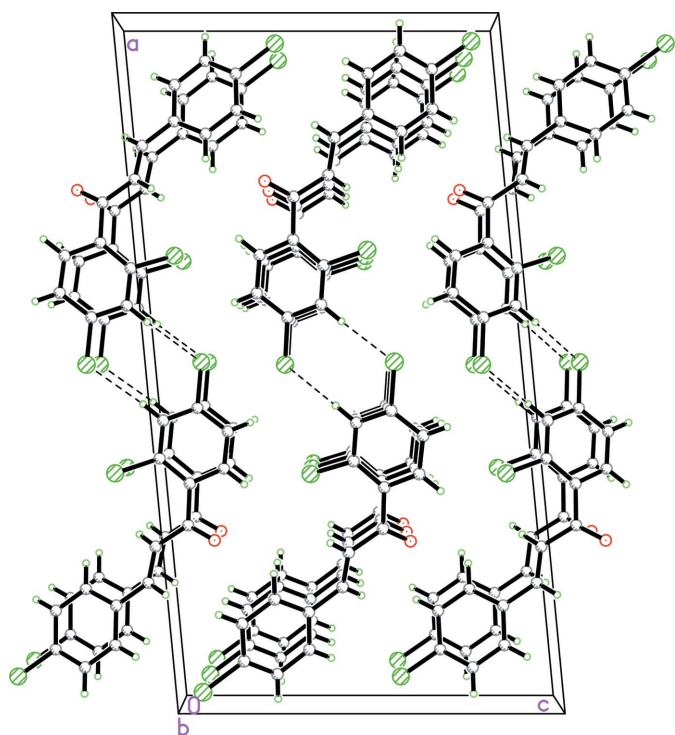
Among many organic compounds reported for their second harmonic generation, chalcone derivatives have attracted much interest as they exhibit extremely high and fast non-linearity (Fichou *et al.*, 1988; Patil *et al.*, 2006; Uchida *et al.*, 1998), much better than that observed in inorganic crystals. In this paper, we report the structure of the title compound, (I). These crystals do not exhibit second-order NLO properties as they crystallize in a centrosymmetric space group.

The bond lengths and angles are within normal ranges (Allen *et al.*, 1987) and compare well with those reported in other related structures (Jeyabharathi *et al.*, 2002; Ng *et al.*, 2006*a,b*; Patil *et al.*, 2006; Ravishankar *et al.*, 2005; Sathiyamoorthi, Chinnakali, Nanjundan, Radhika *et al.*, 2005; Sathiyamoorthi, Chinnakali, Nanjundan, Santhi & Fun, 2005; Sathiyamoorthi, Chinnakali, Nanjundan, Selvam *et al.*, 2005; Sathiyamoorthi, Chinnakali, Nanjundan, Unnithan *et al.*, 2005; Teh *et al.*, 2006). The short  $\text{Cl}2\cdots\text{H}8\text{A}$  (2.82 Å) contact causes the bond angles  $\text{C}5-\text{C}6-\text{C}7$  [124.82 (14)°] and  $\text{C}6-\text{C}7-\text{C}8$  [117.73 (14)°] to deviate significantly from 120°. Furthermore, the short  $\text{H}8\text{A}\cdots\text{H}11\text{A}$  (2.19 Å) contact results in widening of the  $\text{C}9-\text{C}10-\text{C}11$  angle to 122.02 (15)°.

The enone unit  $\text{O}1/\text{C}7-\text{C}9$  and the two benzene rings  $\text{C}1-\text{C}6$  and  $\text{C}10-\text{C}15$  of the chalcone are each planar, with maximum deviations of 0.050 (2), 0.009 (2) and 0.004 (2) Å for atoms C7, C6 and C15, respectively. The molecule is slightly twisted about the  $\text{C}6-\text{C}7$  bond; relevant torsion angles are  $\text{C}1-\text{C}6-\text{C}7-\text{C}8 = -134.30$  (17)°,  $\text{C}6-\text{C}7-\text{C}8-\text{C}9 = -169.94$  (17)°,  $\text{C}7-\text{C}8-\text{C}9-\text{C}10 = 178.00$  (17)° and  $\text{C}8-\text{C}9-\text{C}10-\text{C}15 = 170.03$  (18)°. The dihedral angle between the two benzene rings is 41.99 (5)°. The least-squares plane through the enone unit makes dihedral angles of 44.81 (6) and 5.91 (8)° with the  $\text{C}1-\text{C}6$  and  $\text{C}10-\text{C}15$  benzene rings, respectively.



**Figure 1**  
The structure of (I), showing 50% probability displacement ellipsoids and the atomic numbering.



**Figure 2**  
The crystal packing of (I), viewed down the *b* axis. Intermolecular C—H···Cl interactions are shown as dashed lines.

A very weak intermolecular C—H···Cl interaction is observed in the crystal structure between atoms C4 and Cl1 [ $C4-H4A = 0.93 \text{ \AA}$ ,  $C4 \cdots Cl1 = 3.706 (2) \text{ \AA}$ ,  $H4A \cdots Cl1 = 2.87 \text{ \AA}$  and  $C4-H4A \cdots Cl1 = 151^\circ$ ]. These weak interactions link the molecules, forming wave-like infinite chains along the *a* axis. These chains are stacked parallel to the *b* axis.

## Experimental

Commercially available AR grade 4-chlorobenzaldehyde, 2,4-dichloroacetophenone and ethanol (99%) were used without further purification to synthesize the title chalcone derivative, (I), by adapting the Claisen–Schmidt condensation reaction (Dhar, 1981). 4-Chlorobenzaldehyde (0.01 mol) in ethanol (30 ml) and 4-dichloroacetophenone (0.01 mol) in ethanol (30 ml) were mixed together, and the mixture was treated with an aqueous solution of NaOH (3 ml, 30%). After stirring for 3 h, the contents of the flask were poured into

ice-cold water (250 ml). The resulting crude solid was collected by filtration. The compound was dried and purified by repeated recrystallization from acetone. The purity of the compound was checked by thin layer chromatography. Crystals of (I) suitable for a single-crystal X-ray diffraction study were grown over a period of 12 d by slow evaporation of an acetone solution.

## Crystal data

$C_{15}H_9Cl_3O$   
 $M_r = 311.57$   
Monoclinic,  $P2_1/c$   
 $a = 24.9624 (7) \text{ \AA}$   
 $b = 3.8201 (1) \text{ \AA}$   
 $c = 13.7230 (4) \text{ \AA}$   
 $\beta = 95.430 (1)^\circ$   
 $V = 1302.74 (6) \text{ \AA}^3$   
 $Z = 4$

$D_x = 1.589 \text{ Mg m}^{-3}$   
Mo  $K\alpha$  radiation  
Cell parameters from 5105 reflections  
 $\theta = 0.8\text{--}35.9^\circ$   
 $\mu = 0.69 \text{ mm}^{-1}$   
 $T = 100.0 (1) \text{ K}$   
Block, colourless  
 $0.32 \times 0.15 \times 0.15 \text{ mm}$

## Data collection

Bruker SMART APEX2 CCD area-detector diffractometer  
 $\omega$  scans  
Absorption correction: multi-scan (SADABS; Bruker, 2005)  
 $T_{\min} = 0.777$ ,  $T_{\max} = 0.906$   
26316 measured reflections

6146 independent reflections  
4428 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.053$   
 $\theta_{\max} = 35.9^\circ$   
 $h = -41 \rightarrow 41$   
 $k = -6 \rightarrow 6$   
 $l = -22 \rightarrow 21$

## Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.049$   
 $wR(F^2) = 0.137$   
 $S = 1.07$   
6146 reflections  
172 parameters  
H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0635P)^2 + 0.5011P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.48 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.64 \text{ e \AA}^{-3}$

H atoms were placed in calculated positions and refined as riding, with  $C-H = 0.93 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{carrier atom})$ .

Data collection: APEX2 (Bruker, 2005); cell refinement: APEX2; data reduction: SAINT (Bruker, 2005); program(s) used to solve structure: SHELXTL (Sheldrick, 1998); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

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